Department of Industrial Chemistry

IC (209)312	Silicate Science II	3(2/2-1/3)
Abbreviation		
Prerequisite	IC 311	

This course is opened for MAJOR ELECTIVE COURSE (PLAN I), MAJOR COMPULSORY COURSE (PLAN II)

Course Description

Ceramic phase diagram, phase transformation, solid-solid reactions, drying behavior, silicate reactions, vitrification, and grain growth.

Objectives

- 1. Students will be able to understand thermal effects on reactions and behaviors of silicates.
- 2. Students will be experienced with experiments involving thermal effects on reactions and behaviors of silicates.

Course Content	Lectu	re Hours
1. Ceramic phase diagram		6
2. Phase transformation		3
3. Solid-solid reactions		3
4. Drying behavior		3
5. Silicate reactions		5
6. Vitrification		5
7. Grain growth		5
	Total	30

Laboratory Topics	Laboratory hours
1. Calculation of ceramic body composition from phase diagram	3
2. Calculation of ceramic glaze composition from phase diagram	3
3. Calculation of ceramic pigment composition from phase diagram	3
4. Calculation of cement composition from phase diagram	3
5. Plaster mold making and slip casting	6
6. Mechanical and thermal de-watering	6
7. Ceramic body and glaze firing	3
8. Ceramic pigment firing	3
9. Glass melting	9
10. Cement firing	6
	45

Faculty of Science