Department of Industrial Chemistry

Faculty of Science

IC (209)201 Chemical Stoichiometry 3(3/3-0/0)

Abbreviation

Prerequisite CHEM 104 and CHEM 108 and MATH 104

This course is opened for CORE COURSE

Course Description

Mass balance of non-chemical reaction system, mass balance of chemical reaction system, energy balance of flow system, heat balance of non-chemical reaction system, heat balance of chemical reaction system, combustion, stoichiometry of unit operations and applications of stoichiometry, mass balance and energy balance with problems in chemical industries.

Objectives

- 1. Students will be able to understand the basic principles of chemical stoichiometry, mass balance and energy balance.
- 2. Students will be able to apply the principles of chemical stoichiometry, mass balance and energy balance for analyzing data from chemical reactions and processes.

Course Content	Lecture Hours
1. Introduction	3
2. Mass balance of non-chemical reaction system	5
3. Mass balance of chemical reaction system	5
4. Energy balance of flow system	5
5. Heat balance of non-chemical reaction system	5
6. Heat balance of chemical reaction system	5
7. Combustion	5
8. Stoichiometry of unit operations	5
9. Applications of stoichiometry, mass balance and energy balance with problems	7
in chemical industries	
Total	45